

Ternary Erbium Chromium Sulfides: Structural Relationships and Magnetic Properties

Paz Vaqueiro,[†] Iwona Szkoda,[†] Rodolfo D. Sanchez,[‡] and Anthony V. Powell^{*†}

Department of Chemistry, Heriot-Watt University, Edinburgh EH14 4AS, U.K., and Centro Atómico Bariloche, 8400 Bariloche (RN), Argentina

Received August 7, 2008

Single crystals of four erbium–chromium sulfides have been grown by chemical vapor transport using iodine as the transporting agent. Single-crystal X-ray diffraction reveals that in Er_3CrS_6 octahedral sites are occupied exclusively by Cr^{3+} cations, leading to one-dimensional CrS_4^{5-} chains of edge-sharing octahedra, while in Er_2CrS_4 , Er^{3+} , and Cr^{2+} cations occupy the available octahedral sites in an ordered manner. By contrast, in $\text{Er}_6\text{Cr}_2\text{S}_{11}$ and Er_4CrS_7 , Er^{3+} and Cr^{2+} ions are disordered over the octahedral sites. In Er_2CrS_4 , $\text{Er}_6\text{Cr}_2\text{S}_{11}$, and Er_4CrS_7 , the network of octahedra generates an anionic framework constructed from M_2S_5 slabs of varying thickness, linked by one-dimensional octahedral chains. This suggests that these three phases belong to a series in which the anionic framework may be described by the general formula $[\text{M}_{2n+1}\text{S}_{4n+3}]^{x-}$, with charge balancing provided by Er^{3+} cations located in sites of high-coordination number within one-dimensional channels defined by the framework. Er_4CrS_7 , $\text{Er}_6\text{Cr}_2\text{S}_{11}$, and Er_2CrS_4 may thus be considered as the $n = 1, 2$, and ∞ members of this series. While Er_4CrS_7 is paramagnetic, successive magnetic transitions associated with ordering of the chromium and erbium sub-lattices are observed on cooling Er_3CrS_6 ($T_{\text{C}}(\text{Cr}) = 30$ K; $T_{\text{C}}(\text{Er}) = 11$ K) and Er_2CrS_4 ($T_{\text{N}}(\text{Cr}) = 42$ K, $T_{\text{N}}(\text{Er}) = 10$ K) whereas $\text{Er}_6\text{Cr}_2\text{S}_{11}$ exhibits ordering of the chromium sub-lattice only ($T_{\text{N}} = 11.4$ K).

Introduction

Mixed-metal chalcogenides containing transition-series elements, either exclusively or in conjunction with a diamagnetic main-group cation have been extensively studied for their structural, magnetic, and electronic properties.^{1,2} In contrast, chalcogenides containing a transition-series element in conjunction with a rare-earth cation represent something of an under-developed area in solid-state chemistry. The high and variable coordination number of a rare-earth cation, offers a degree of structural complexity that goes beyond that in materials containing main-group or transition-metal ions with well-defined coordination preferences. Furthermore, the presence of d- and f-block elements allows three types of electron-electron interaction, *d-d*, *d-f*

and *f-f*, to be simultaneously present affording the potential for a corresponding richness in the magnetic and transport properties.

The comparatively small number of examples of rare-earth transition-metal chalcogenides reported to date is dominated by the sulfides, with considerably fewer examples of selenides and only a handful of tellurides known.³ Moreover, even for the simplest ternary sulfides, systematic studies for a given stoichiometry of the structure and physical properties as a function of the rare-earth and/or transition-metal cation remain to be performed. For example, while the synthesis and lattice parameters of the series of phases Ln_3CrS_6 ($\text{Ln} = \text{La}–\text{Lu}$)⁴ have been reported, crystal structures have not been determined and little is known about their physical properties.

In many cases, structure has been inferred solely from comparison of powder X-ray diffraction data with those of known phases and detailed structural investigations, including the nature of any cation ordering, have not been carried out.

* To whom correspondence should be addressed. E-mail: a.v.powell@hw.ac.uk. Fax: +44 (0)131 451 3180.

[†] Heriot-Watt University.

[‡] Centro Atómico Bariloche.

(1) Rouxel, J. In *Physics and Chemistry of Low-Dimensional Inorganic Conductors* Schlenker, C.; Dumas, J.; Greenblatt, M.; van Smaalen, S., Eds.; NATO ASI Series B, 1996; Vol. 354, p 1.

(2) Pearce, C.I.; Patrick, R. A. D.; Vaughan, D. J. *Rev. Mineral. Geochem.* **2006**, *61*, 127.

(3) Mitchell, K.; Ibers, J. A. *Chem. Rev.* **2002**, *102*, 1929.

(4) Rustamov, P.G.; Kurbanov, T. Kh.; Aliev, O. M.; Aliev, I. P. *Inorg. Mater.* **1984**, *20*, 1664.

For example, single crystal X-ray diffraction data obtained for Y_4FeS_7 and Ho_4FeS_7 indicate a Y_5S_7 -type structure^{5,6} and, on the basis of powder X-ray diffraction data, it was suggested that other materials of general formula Ln_4MS_7 ($\text{Ln} = \text{Y}, \text{Dy}, \text{Ho}, \text{Er}, \text{Tm}, \text{Yb}$; $\text{M} = \text{Mg}, \text{Cr}, \text{Mn}, \text{Fe}, \text{Cd}$) are isostructural. However, it appears that related materials, initially formulated Ln_4MS_7 ($\text{M} = \text{Mn}, \text{Fe}, \text{Co}, \text{Ni}$; $\text{Ln} = \text{La}, \text{Ce}, \text{Pr}$ and Nd),⁷ are more complex than originally proposed since analysis of single-crystal X-ray diffraction data for “ Ce_4FeS_7 ” reveals a different structure and a crystallographically-determined composition $\text{Ce}_3\text{Fe}_{1.94}\text{S}_7$.⁸

On the basis of powder X-ray diffraction data, three different structure types have been proposed for materials of general formula, Ln_2MS_4 ($\text{M} = \text{Mg}, \text{Cr}, \text{Mn}, \text{Fe}$), depending on the identity of the rare-earth cation.⁹ For the early lanthanides ($\text{Ln} = \text{La} - \text{Gd}$), an orthorhombic phase was identified, while rare-earth ions drawn from later in the series ($\text{Tb} - \text{Tm}$) generally favor adoption of a Y_2MnS_4 -type structure and a spinel-type structure is also observed for Yb-containing phases. However, recent single-crystal X-ray diffraction studies indicate greater complexity and reveal that the phases of nominal stoichiometry “ La_2CrS_4 ” and “ La_2FeS_4 ” are rhombohedral and better formulated $\text{La}_{15.9}\text{Cr}_{5.4}\text{S}_{32}$ ¹⁰ and $\text{La}_{52}\text{Fe}_{12}\text{S}_{90}$ ¹¹ respectively. Single-crystal X-ray diffraction studies of Y_2MnS_4 -type materials are confined to “ Yb_2MnS_4 ”¹² and Er_2CrS_4 .¹³ In the former this led to its reformulation as $\text{Yb}_{2.33}\text{Mn}_{0.67}\text{S}_4$, while an enlarged unit cell was observed in the latter. Examination by electron microscopy of materials within the $\text{MnS} - \text{Er}_2\text{S}_3$ ¹⁴ system has confirmed the existence of Y_2MnS_4 -type and Y_5S_7 -type phases, and suggested that other phases with intermediate compositions such as $\text{Er}_6\text{Mn}_2\text{S}_{11}$ and $\text{Er}_3\text{Mn}_3\text{S}_{15}$ may exist. While a number of rare-earth transition-metal sulfides of general formula $\text{Ln}_6\text{M}_2\text{S}_{11}$ have been reported,^{14,15} detailed structural analysis has been performed only for $\text{Yb}_6\text{Fe}_2\text{S}_{11}$ ¹⁶ and $\text{Er}_6\text{Cr}_2\text{S}_{11}$.¹⁷

In this work we have sought to carry out a systematic study of the structural and magnetic properties of ternary erbium–chromium sulfides. The presence, or otherwise, of long-range magnetic order at low temperatures is correlated with the detailed structures determined by single-crystal

X-ray diffraction. Moreover, the identification of a common building block leads us to propose that these materials be considered the first members of a homologous series of compounds.

Experimental Section

Bulk powdered samples of all materials were prepared from mixtures of high-purity erbium (Sigma-Aldrich, 99.9%), chromium (Alfa Aesar, 99.95%) and sulfur (Sigma Aldrich, 99.99%) powders of appropriate stoichiometry. The elements were ground in an agate mortar inside an argon-filled glovebox prior to sealing into evacuated ($< 10^{-4}$ Torr) silica tubes, together with 10 mg of iodine. Er_3CrS_6 , Er_4CrS_7 , and $\text{Er}_6\text{Cr}_2\text{S}_{11}$ were prepared by heating the reaction mixtures at 1273 K for 1 week, whereas synthesis of Er_2CrS_4 required 2 weeks at this temperature. All products were cooled to room temperature at the natural rate of the furnace (*ca.* 5 K min^{-1}).

Single crystals of the four ternary sulfides were grown by chemical vapor transport from stoichiometric mixtures of the elements, using iodine (3–5 mg cm^{-3}) as the transporting agent. The charge and growth zones were held at 1273 K and 1223 K respectively for the growth of Er_3CrS_6 and $\text{Er}_6\text{Cr}_2\text{S}_{11}$ crystals, while a steeper gradient (1273 K to 1173 K) was required for the formation of single crystals of Er_2CrS_4 and Er_4CrS_7 . In all cases crystals suitable for structure determination by single-crystal X-ray diffraction were obtained in 7 days.

Powder X-ray diffraction data for bulk samples were obtained using a Bruker D8 Advance powder diffractometer, operating with germanium-monochromated $\text{Cu K}\alpha_1$ radiation ($\lambda = 1.5405 \text{ \AA}$) and fitted with a Bruker LynxEye linear detector. Data were collected over the angular range $10 \leq 2\theta^\circ \leq 120$ in 0.014° angular increments, counting for 2.4 s at each step. Magnetic measurements over the temperature range $90 \leq T/\text{K} \leq 300$ were performed using a LakeShore 7300 series vibrating-sample magnetometer (VSM). Samples were contained in Lucite capsules and measurements were conducted in an applied field of 1000 G, after cooling in zero applied field (*zfc*). Measurements were extended to lower temperatures ($5 \leq T/\text{K} \leq 150$) using a Quantum Design MPMS-5S SQUID susceptometer. Samples were loaded into gelatin capsules at room temperature, and data were collected in a measuring field of 1000 G after cooling under *zfc* conditions. In cases where magnetic transitions were identified, low-temperature magnetic susceptibilities were also measured after cooling in the applied field. Magnetization data as a function of field were collected in the range $0 \leq H/\text{G} \leq 10000$. Data were corrected for the diamagnetism of the gelatin capsule and for intrinsic core diamagnetism.

Single crystals were mounted on glass fibers, and X-ray intensity data collected using a Bruker X8 APEX 2 diffractometer¹⁸ with graphite monochromated $\text{Mo K}\alpha$ radiation ($\lambda = 0.71073 \text{ \AA}$). Full crystallographic details are given in Table 1. The structures were solved by direct methods using the program SIR92.¹⁹ Subsequent Fourier calculations and least-squares refinements on F were carried out in the CRYSTALS program suite.²⁰ A Chebyshev polynomial was applied as a weighting scheme.²¹ Refined parameters are

- (5) Adolphe, C. *Ann. Chim.* **1965**, *10*, 271.
- (6) Adolphe, C.; Laruelle, P. *Bull. Soc. Fr. Mineral. Cristallogr.* **1968**, *91*, 219.
- (7) Collin, G.; Rouyer, F.; Loriers, J. *C.R. Acad. Sci. Paris, Series C* **1968**, *266*, 689.
- (8) Mills, A. M.; Ruck, M. *Acta Crystallogr. C* **2004**, *60*, i71.
- (9) Patrie, M.; Chevalier, R. *C.R. Acad. Sci. Paris, Series C* **1966**, *263*, 1061.
- (10) Litteer, J. B.; Sirchio, S. A.; Fettingner, J. C.; Smolyaninova, V.; Eichhorn, B. W.; Greene, R. L. *Chem. Mater.* **1999**, *11*, 1179.
- (11) Mills, A. M.; Bräunling, D.; Ruck, M. *Acta Crystallogr. C* **2006**, *62*, i70.
- (12) Lemoine, P. P.; Tomas, A.; Guittard, M. *Acta Crystallogr. C* **1992**, *48*, 774.
- (13) Tomas, A.; Chevalier, R.; Laurelle, P.; Bachet, B. *Acta Crystallogr. B* **1976**, *32*, 3287.
- (14) Landa-Canovas, A. R.; Otero-Diaz, L. C. *Aust. J. Chem.* **1992**, *45*, 1473.
- (15) Bakker, M.; Hyde, B. G. *Philos. Mag. A* **1978**, *38*, 615.
- (16) Chaqour, S. M.; Tomas, A.; Lemoine, P.; Palazzi, M.; Guittard, M. *Acta Crystallogr. C* **1994**, *50*, 1655.
- (17) Tomas, A.; Rigoult, J.; Guittard, M.; Laruelle, P. *Acta Crystallogr. B* **1980**, *36*, 1987.

- (18) Bruker X8 APEX 2, Version 1.0-8; Bruker AXS Inc: Madison, WI, 2003.
- (19) Altomare, A.; Cascarano, G.; Giacovazzo, C.; Guagliardi, A.; Burla, M. C.; Polidori, G.; Camelli, M. *J. Appl. Crystallogr., Sect. A* **1994**, *27*, 435.
- (20) Watkin, D. J.; Prout, C. K.; Carruthers, J. R.; Betteridge, P. W. In *CRYSTALS Issue 10*: Chemical Crystallography Laboratory, University of Oxford: U.K., 1996.
- (21) Carruthers, J. R.; Watkin, D. J. *Acta Crystallogr. A* **1979**, *35*, 698.

Table 1. Crystallographic Data for Ternary Erbium Chromium Sulfides Investigated by Single-Crystal X-ray Diffraction

Formula	Er ₃ CrS ₆	Er ₂ CrS ₄	Er ₆ Cr ₂ S ₁₁	Er ₄ CrS ₇
M_r	746.17	514.78	1462.48	940.93
crystal habit	black needle	black needle	black needle	black block
dimensions/mm	0.5 × 0.04 × 0.04	0.2 × 0.04 × 0.02	0.26 × 0.08 × 0.06	0.16 × 0.08 × 0.05
crystal system	orthorhombic	orthorhombic	orthorhombic	monoclinic
space group	<i>Pnmm</i>	<i>P2₁ca</i>	<i>Cmcm</i>	<i>C2/m</i>
T/K	293	293	293	293
$a/\text{Å}$	13.192(2)	12.4629(11)	3.7551(3)	12.5057(5)
$b/\text{Å}$	15.819(3)	7.4882(6)	12.5190(8)	3.7650(2)
$c/\text{Å}$	3.7333(6)	12.4937(11)	34.177(2)	11.2740(4)
β/deg	90.0	90.0	90.0	105.85(2)
$V/\text{Å}^3$	779.1(2)	1165.97(17)	1606.7(2)	510.64(4)
Z	4	8	4	2
wavelength/Å	0.71073	0.71073	0.71073	0.71073
μ/mm^{-1}	34.848	31.636	33.749	34.618
measured data	17332	3248	18621	6925
unique data	2087	3248	1410	878
observed data ($I > 3\sigma(I)$)	1729	1528	1086	735
R_{merg}	0.0371	0.0320	0.0484	0.0313
$R(F_o)^a$	0.0186	0.0328	0.0229	0.0332
$R_w(F_o)^b$	0.0188	0.0128	0.0235	0.0279

$$^a R(F_o) = \frac{\sum ||F_o| - |F_c||}{\sum |F_o|}, \quad ^b R_w(F_o) = \frac{[\sum w(|F_o| - |F_c|)^2 / \sum w|F_o|^2]^{1/2}}{}$$

Table 2. Final Refined Coordinates and Equivalent Isotropic Displacement Parameters for Er₃CrS₆

atom	site	x	y	z	$U_{\text{eq}}(\text{Å}^2)$
Er(1)	4(g)	0.54922(1)	0.22090(1)	0	0.0042(1)
Er(2)	4(g)	0.31855(1)	0.35227(1)	0	0.0041(1)
Er(3)	4(g)	0.25639(1)	0.09874(1)	-1/2	0.0048(1)
Cr(4)	2(c)	1/2	1/2	-1/2	0.0033(2)
Cr(5)	2(c)	1/2	0	-1/2	0.0028(2)
S(6)	4(g)	0.41659(8)	0.07611(7)	0	0.0051(2)
S(7)	4(g)	0.62050(8)	0.10867(7)	-1/2	0.0052(2)
S(8)	4(g)	0.68796(8)	0.28959(7)	1/2	0.0050(2)
S(9)	4(g)	0.39569(8)	0.24634(7)	-1/2	0.0048(2)
S(10)	4(g)	0.52197(8)	0.40137(7)	0	0.0051(2)
S(11)	4(g)	0.31847(8)	0.47972(7)	1/2	0.0051(2)

Table 3. Final Refined Coordinates and Equivalent Isotropic Displacement Parameters for Er₂CrS₄

atom	site	x	y	z	$U_{\text{eq}}(\text{Å}^2)$
Er(1)	4(a)	0.58027(12)	0.6365(3)	1.13561(14)	0.0098(4)
Er(2)	4(a)	0.57184(11)	1.1359(3)	1.13675(15)	0.0116(4)
Er(3)	4(a)	0.7458(2)	0.8843(3)	1.37394(9)	0.0094(3)
Er(4)	4(a)	0.8845(2)	0.6139(3)	1.61095(9)	0.0094(3)
Cr(5)	4(a)	0.8792(8)	1.1153(11)	1.6094(3)	0.0105(9)
Cr(6)	4(a)	0.7495(7)	1.3822(12)	1.3771(3)	0.0098(8)
S(7)	4(a)	0.5498(4)	0.8824(17)	1.2896(5)	0.0086(11)
S(8)	4(a)	0.9380(5)	0.8836(14)	1.4728(4)	0.0096(10)
S(9)	4(a)	0.4322(5)	0.3878(13)	1.0477(4)	0.0087(10)
S(10)	4(a)	0.6933(5)	0.6167(15)	1.5128(5)	0.0086(10)
S(11)	4(a)	0.7984(6)	1.1502(8)	1.2453(5)	0.0120(10)
S(12)	4(a)	0.7990(5)	0.6476(8)	1.2263(5)	0.0100(10)
S(13)	4(a)	0.7004(5)	1.1151(13)	1.5305(4)	0.0088(10)
S(14)	4(a)	0.5735(5)	0.3902(16)	1.2921(5)	0.0094(10)

presented in Tables 2–5 and selected bond distances are provided as Supporting Information.

Results

Structural Investigations. The asymmetric unit of Er₃CrS₆ contains two and three crystallographically-distinct chromium and erbium ions, respectively. Both chromium ions are trivalent and exhibit a slightly distorted octahedral coordination, with an average Cr–S distance of 2.44 Å. In contrast, two different coordination environments may be identified for the rare-earth cations. Er(1) and Er(3) are surrounded by eight sulfur atoms in a bicapped trigonal-prismatic coordination, while Er(2) is seven coordinate and adopts a capped pseudo-octahedral arrangement. The average

Er–S distances, which increase with increasing coordination number, are similar to those in the binary sulfide, Er₂S₃,²² in which Er³⁺ cations exhibit coordination numbers in the range six to eight.

Single-crystal X-ray diffraction data for Er₂CrS₄ were initially indexed in the centrosymmetric space group *Pbca*. However, detailed examination of these data revealed a number of weak (*0kl*) reflections which violate the $k = 2n$ condition of the *b* glide plane. The structure was subsequently solved in the non-centrosymmetric space group *P2₁ca*, as previously proposed by Tomas et al.¹³ With the exception of S(11) and S(12), the structure exhibits pseudosymmetry,

(22) Fang, C. M.; Meetsma, A.; Wiegers, G. A.; Boom, G. *J. Alloys Compd.* **1993**, *201*, 255.

Table 4. Final Refined Coordinates and Equivalent Isotropic Displacement Parameters for Er₆Cr₂S₁₁

atom	site	x	y	z	U _{eq} (Å ²)	SOF
Er(1)	4(c)	0	0.49117(3)	1/4	0.0159(1)	
Er(2)	8(f)	1/2	0.75959(2)	0.43838(1)	0.0093(1)	
M(3)	8(f)	0	0.74867(3)	0.31710(1)	0.0093(1)	Er:0.583(3) Cr:0.417(3)
M(4)	8(f)	1/2	0.51374(3)	0.36581(1)	0.0086(1)	Er:0.569(3) Cr:0.431(3)
M(5)	4(c)	1/2	1/2	1/2	0.0071(1)	Er:0.714(4) Cr:0.286(4)
S(6)	8(f)	1/2	0.60363(14)	0.29733(5)	0.0146(4)	
S(7)	4(c)	1/2	0.33818(17)	1/4	0.0154(6)	
S(8)	8(f)	0	0.65877(13)	0.38691(5)	0.0122(4)	
S(9)	8(f)	1/2	0.41348(12)	0.43103(5)	0.0119(4)	
S(10)	8(f)	0	0.64706(10)	0.48397(4)	0.0068(3)	
S(11)	8(f)	1/2	0.88752(17)	0.33672(8)	0.0303(6)	

Table 5. Final Refined Coordinates and Equivalent Isotropic Displacement Parameters for Er₄CrS₇

atom	site	x	y	z	U _{eq} (Å ²)	SOF
Er(1)	4(i)	0.30774(3)	0	0.19747(3)	0.0129(1)	
M(2)	4(i)	1/2	1/2	0	0.0103(2)	Er: 0.763(6) Cr: 0.237(6)
M(3)	4(i)	0.61409(4)	1/2	0.42473(4)	0.0106(1)	Er: 0.599(4) Cr: 0.401(4)
S(4)	4(i)	0.16018(15)	1/2	0.05072(17)	0.0100(4)	
S(5)	4(i)	0.46519(16)	1/2	0.21708(19)	0.0149(5)	
S(6)	4(i)	0.75938(17)	1/2	0.64334(18)	0.0150(5)	
S(7)	4(i)	0.4861(4)	0	0.4814(4)	0.0184(8)	0.5(–)

Table 6. Lattice Parameters of Bulk Samples of Erbium–Chromium Sulfides Determined by Powder X-ray Diffraction

	a/Å	b/Å	c/Å	β/°
Er ₃ CrS ₆	13.1815(4)	15.8363(5)	3.7293(2)	90.0(–)
Er ₂ CrS ₄	12.4463(5)	7.4892(3)	12.4839(6)	90.0(–)
Er ₆ Cr ₂ S ₁₁	3.7568(3)	12.513(1)	34.121(3)	90.0(–)
Er ₄ CrS ₇	12.5007(9)	3.7602(3)	11.2637(8)	105.977(5)

and therefore similarity restraints were applied to the anisotropic thermal parameters during the least-squares refinements.

There are two crystallographically-distinct chromium and four erbium ions in the asymmetric unit of Er₂CrS₄. All chromium ions exhibit a distorted octahedral coordination, with two long and four short Cr–S distances, of *ca.* 2.84 and 2.48 Å respectively. This appears to be a rare example of a static Jahn–Teller distortion associated with the degenerate ground state of the Cr²⁺:d⁴ ion, whose presence is required by formal valence considerations and supported by bond valence²³ sums. Although Jahn–Teller distortions of Cr²⁺ have been reported in halides such as Rb₂CrCl₄ and ACrCl₃ (A = Cs, Rb),^{24,25} the binary phase CrS represents one of the few previously characterized examples of such a distortion in a sulfide.²⁶ The erbium ions exhibit two different coordination environments, octahedral and monocapped trigonal prismatic, with average Er–S bond lengths of *ca.* 2.67 Å and *ca.* 2.82 Å respectively.

Er₆Cr₂S₁₁ has been reported to crystallize in the non-centrosymmetric space group *Bb2₁m*, an alternative setting of *Cmc2₁*.¹⁷ However, application of Platon/Addsym^{27,28} to

identify missing symmetry elements indicates that there is an additional mirror plane, and that the structure is more correctly described in the centrosymmetric space group *Cmcm*. Refinements using single-crystal data were therefore carried out in *Cmcm*, and resulted in a low weighted residual of 2.35% (Table 1). The asymmetric unit contains three crystallographically-independent cation sites of octahedral coordination, together with two erbium sites of distorted trigonal prismatic and monocapped trigonal prismatic coordination. The Cr²⁺ and Er³⁺ cations are disordered over the three octahedral sites. The average metal-sulfur distance at the octahedral sites is *ca.* 2.65 Å which is slightly larger than that observed for divalent chromium in Er₂CrS₄ and is consistent with the presence of both chromium and the larger erbium cations at these sites.

There are three crystallographically-independent cation sites in Er₄CrS₇, two of which are octahedrally coordinated by sulfur while the third is a seven-coordinate monocapped trigonal-prismatic site. Structural refinement using single-crystal X-ray diffraction data reveal that erbium and chromium ions are disordered over both octahedral sites, with the remaining erbium ions occupying the seven-coordinate site. The average cation-sulfur distances at the octahedral sites is *ca.* 2.66 Å. The average Er–S distance at the seven-coordinated site is *ca.* 2.82 Å, comparable to those found in the structures of Er₂CrS₄ and Er₃CrS₆. During initial structural refinements, the anion S(7) was located at a 2(*d*) (0.5, 0, 0.5) site. This resulted in large oblate anisotropic thermal parameters, suggesting that this ion might be disordered. In subsequent refinements S(7) was placed at a 4(*i*) site (0.4861(4), 0, 0.4814(4)), with a site occupancy factor of 0.5. Similar disorder has previously been reported for the isostructural Tm₅S₇.²⁹

(23) Brese, N. E.; O'Keeffe, M. *Acta Crystallogr. B* **1991**, *47*, 192.(24) Day, P.; Hutchings, M. T.; Janke, E.; Walker, P. J. *J. Chem. Soc., Chem. Commun.* **1979**, 711.(25) Crama, W. J.; Maaskant, W. J. A.; Verschoor, G. C. *Acta Crystallogr. B* **1978**, *34*, 1973.(26) Jellinek, F. *Acta Crystallogr.* **1957**, *10*, 620.(27) Speck, A. L. *J. Appl. Crystallogr.* **2003**, *36*, 7.(28) Speck, A. L. *J. Appl. Crystallogr.* **1988**, *21*, 578.

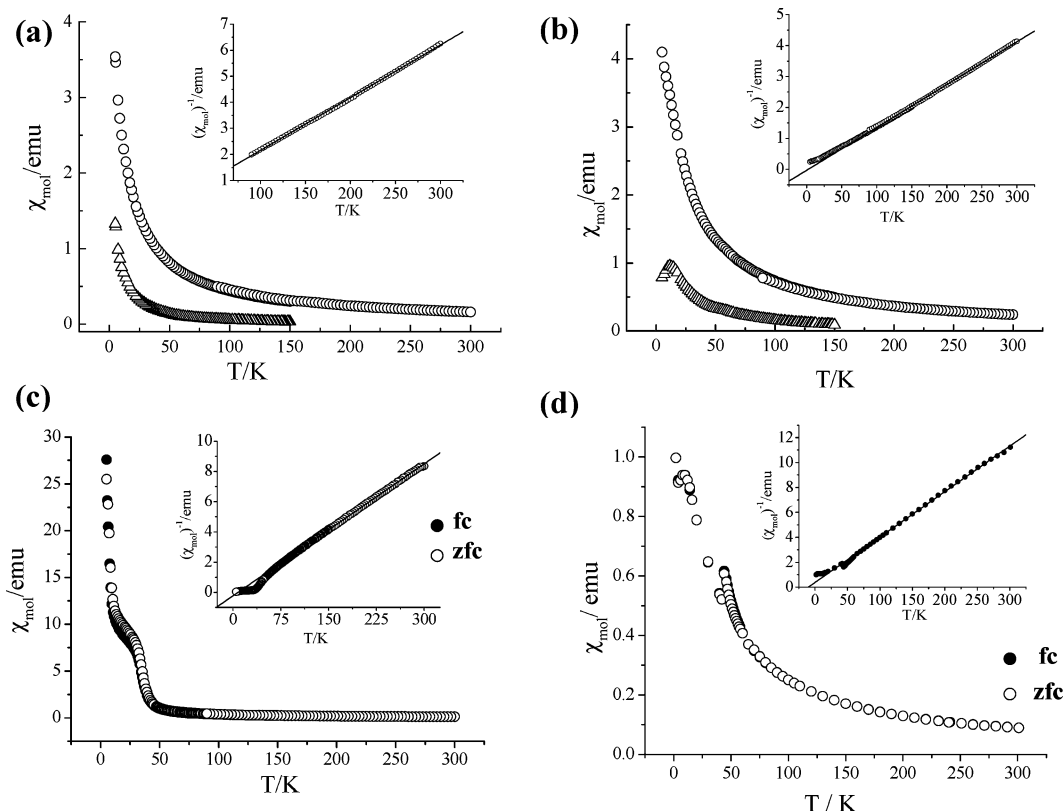


Figure 1. Magnetic susceptibility data for (a) Er_4CrS_7 , (b) $\text{Er}_6\text{Cr}_2\text{S}_{11}$, (c) Er_3CrS_6 , and (d) Er_2CrS_4 . Solid points denote field-cooled data and open points zero-field cooled data. In (a, b), open triangles represent the magnetic susceptibilities obtained following subtraction of the contribution due to the erbium moments. The insets show the fit of a Curie–Weiss expression to high-temperature data collected using a vibrating sample magnetometer.

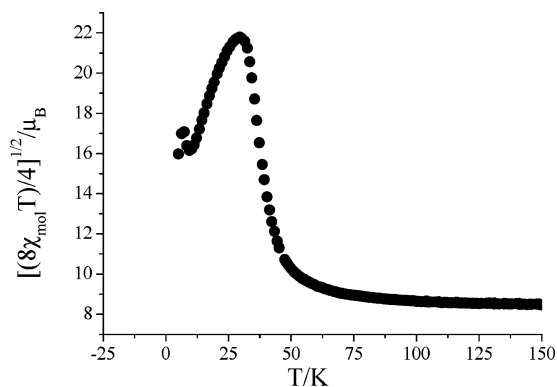


Figure 2. Temperature dependence of the effective magnetic moment per cation of Er_3CrS_6 , as measured by the quantity $(8\chi T/4)^{1/2}$.

Powder X-ray diffraction data for bulk samples prepared by high-temperature synthesis are consistent with the formation of single-phase products. The data can be indexed on the basis of the unit-cells determined by single-crystal X-ray diffraction, with refined parameters that are in good agreement with those obtained in the single-crystal study (Table 6).

Magnetic Properties. Magnetic susceptibility data (Figure 1) reveal that all materials follow Curie–Weiss behavior at high temperatures. The effective magnetic moments, μ_{eff} , (Table 7) derived from the fit of a Curie–Weiss expression to the data obtained by the VSM technique are in good agreement with expectations based on spin-only behavior for

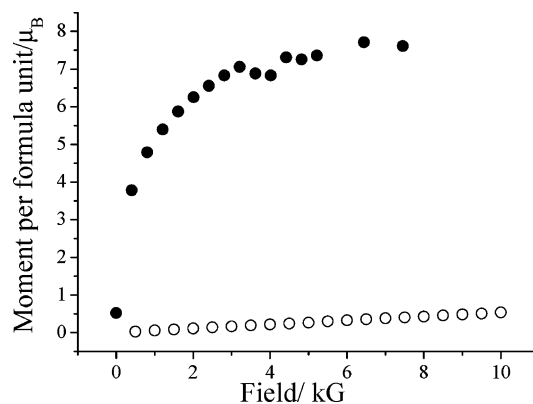


Figure 3. Variation in the magnetic moment per formula unit of Er_3CrS_6 (solid points) and Er_2CrS_4 (open points) as a function of applied magnetic field at 5 K.

the chromium cations and a free-ion value of $9.58 \mu_{\text{B}}$, calculated from the Landé formula, for the μ_{eff} of the Er^{3+} cations. Er_4CrS_7 exhibits a small negative Weiss constant, indicating that the dominant magnetic interactions are antiferromagnetic in origin. However, these interactions are of insufficient strength to produce long-range magnetic order and the material remains paramagnetic to the lowest temperature studied. While the data for $\text{Er}_6\text{Cr}_2\text{S}_{11}$ also appear to be consistent with paramagnetic behavior down to 5 K and lead to a very small positive Weiss constant, a significant deviation from Curie–Weiss behavior is observed below *ca.* 20 K. Subtraction of the contribution to the magnetic susceptibility arising from the erbium ions, assuming Curie–Weiss behavior and an effective magnetic moment

(29) Range, K. J.; Rau, F.; Klement, U. Z. *Kristallogr.* **1993**, *208*, 97.

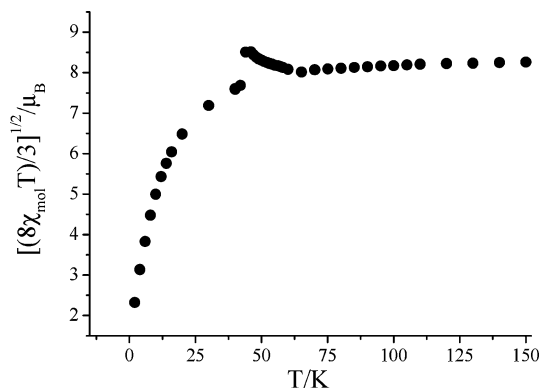


Figure 4. Temperature dependence of the effective magnetic moment per cation of Er_2CrS_4 , as measured by the quantity $(8\chi T/3)^{1/2}$.

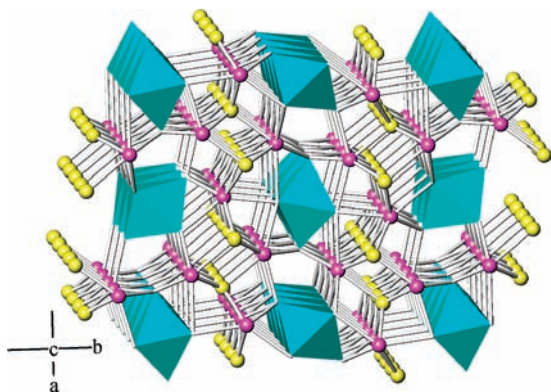


Figure 5. Polyhedral representation of the crystal structure of Er_3CrS_6 viewed along [001]. CrS_6 octahedra are colored cyan, erbium cations are represented by magenta circles, and sulfide anions by yellow circles.

of $9.58 \mu_{\text{B}}$, reveals an underlying antiferromagnetic transition, with $T_{\text{N}} = 11.4 \text{ K}$ (Figure 1b). This suggests ordering of the Cr^{2+} sub-lattice occurs on cooling, the signature of which is masked by the much larger magnetic response of the rare-earth ion. A similar subtraction applied to the data for Er_4CrS_7 does not reveal a corresponding ordering transition of the Cr^{2+} sub-lattice in this phase.

The magnetic susceptibility data for Er_3CrS_6 , although exhibiting Curie–Weiss behavior at high temperatures with a value of μ_{eff} consistent with trivalent rare-earth and transition-metal cations, shows more complex behavior at low temperatures, with slight divergence of *fc* and *zfc* data in the temperature range $5 \leq T/\text{K} \leq 45$. Moreover, on cooling below 50 K, the $\chi(T)$ curve exhibits an inflection at 30 K, followed by a steep rise at 10 K, suggesting two successive magnetic transitions. Further support for this is provided by examination of the temperature dependence of the effective magnetic moment, μ_{eff} , per cation (Figure 2), as measured by the quantity $[(8\chi T)/4]^{1/2}$. On cooling, μ_{eff} remains close to the expected value for Er^{3+} and Cr^{3+} over a wide range of temperature, until at *ca.* 50 K it begins to increase markedly, reaching a maximum of $21.8 \mu_{\text{B}}$ at 30 K. Below this temperature μ_{eff} falls, reaches a minimum at 9.4 K, before increasing slightly on further cooling. Measurement of the magnetization as a function of field confirms the existence of a spontaneous magnetization at 5 K (Figure 3). This approaches saturation in an applied field of 10 kG. The saturated moment was estimated by taking the intercept

on the moment axis of linear fits to the high-field region of μ versus $1/H$ plots, leading to a value of $7.97(9) \mu_{\text{B}}$ per formula unit.

Two magnetic transitions are apparent at low temperature in the magnetic susceptibility data of Er_2CrS_4 . The value of μ_{eff} , extracted from the Curie–Weiss fit to the high-temperature data, is consistent with the presence of Er^{3+} and Cr^{2+} ions. On cooling, the magnetic susceptibility shows a peak at 44 K. Below 42 K, χ rises and exhibits a second broad maximum at 9 K, before rising again below 4 K. The negative Weiss constant indicates that the dominant magnetic exchange interactions are antiferromagnetic in origin. The effective magnetic moment is almost invariant with temperature down to *ca.* 44 K (Figure 4), below which μ_{eff} decreases markedly, suggesting that the peak in $\chi(T)$ at 44 K is associated with long-range antiferromagnetic ordering of the chromium sub-lattice; the second lower-temperature anomaly corresponding to ordering of the erbium sub-lattice. The magnetization shows an almost linear dependence on magnetic field (Figure 3) and does not approach saturation even at the highest field used. This, together with the relatively low value of *ca.* $0.5 \mu_{\text{B}}$ per cation at 10 kG indicates that there is no uncompensated moment associated with either the erbium or chromium sub-lattices.

Discussion

Single-crystal X-ray diffraction reveals that Er_3CrS_6 adopts the U_3ScS_6 -type structure,³⁰ with complete ordering of Cr^{3+} cations into the octahedral sites. It appears likely on the basis of powder X-ray diffraction data that other sulfides of stoichiometry Ln_3CrS_6 ($\text{Ln} = \text{La}–\text{Lu}$)⁴ are isostructural, although determination of the degree of $\text{Ln}^{3+}/\text{Cr}^{3+}$ ordering requires detailed structural measurements. While isostructural phases containing other transition metals have not been reported, it is possible to substitute chromium by indium, to form the closely related Ln_3InS_6 ($\text{Ln} = \text{Sm}$ to Tb) phases.^{31,32} The structure of Er_3CrS_6 consists of $[\text{CrS}_4]^-$ chains of edge-sharing Cr^{3+} -centered octahedra directed along [001] (Figure 5). $\text{Er}(1)\text{S}_8$ polyhedra share opposite triangular faces to form ErS_5^{5-} chains, also directed parallel to the crystallographic *c*-axis. The ErS_5^{5-} chains share common edges and vertices with the CrS_4^{5-} chains and serve to link the latter to form a slab parallel to the (100) plane. These slabs are connected in the [100] direction by $\text{Er}(2)\text{S}_7$ and $\text{Er}(3)\text{S}_8$ one-dimensional chains, to generate a complex three-dimensional network. The erbium-sulfur matrix serves to isolate individual CrS_4^{5-} chains from each other; the closest interchain $\text{Cr}(1)\cdots\text{Cr}(2)$ separation being 6.85 \AA . Although the intra-chain $\text{Cr}–\text{Cr}$ distances are considerably shorter at 3.733 \AA , this is significantly greater than the critical distance ($R_{\text{C}} \approx 3.1 \text{ \AA}$)³³ below which a direct $t_{2g}–t_{2g}$ interaction could occur. Intra-chain magnetic exchange is therefore through a 90° superexchange

(30) Rodier, N.; Tien, V. *Acta Crystallogr. B* **1976**, *32*, 2705.

(31) Messain, D.; Carré, D.; Laurelle, P. *Acta Crystallogr. B* **1977**, *33*, 2540.

(32) Guittard, M.; Carré, D.; Kabré, T. S. *Mater. Res. Bull.* **1978**, *13*, 279.

(33) Goodenough, J.B. In *Magnetism and the Chemical Bond*; Wiley: New York, 1963.

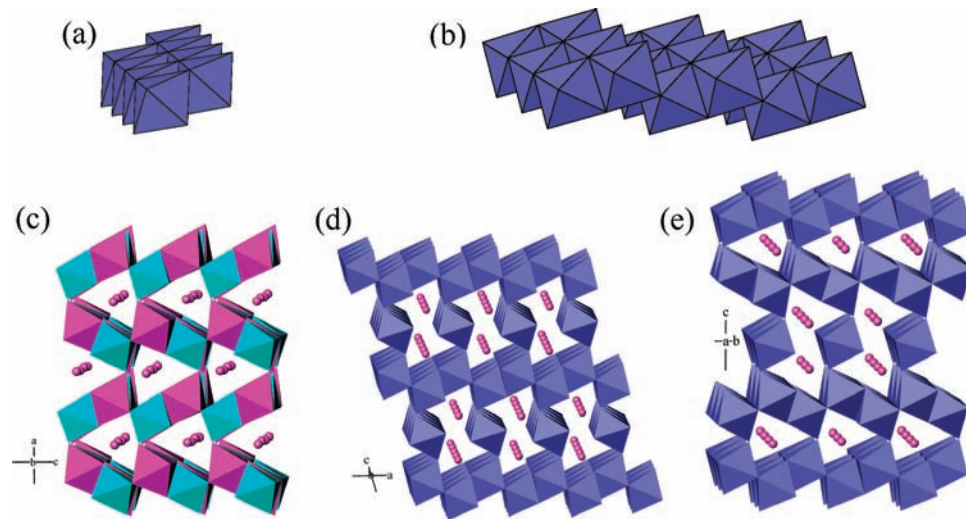


Figure 6. The relationship between the structures of Er_2CrS_4 , Er_4CrS_7 , and $\text{Er}_6\text{Cr}_2\text{S}_{11}$: (a) the double octahedral chain that forms the basic building unit for the structures of these materials; (b) the M_2S_5 slab formed from linkage of double octahedral chains; (c) Er_2CrS_4 viewed along the $[010]$ direction; (d) the structure of Er_4CrS_7 viewed along the $[010]$ direction; (e) $\text{Er}_6\text{Cr}_2\text{S}_{11}$ viewed along the $[100]$ direction. Non-framework erbium cations are colored magenta. Cyan- and magenta-colored octahedra within the cation-ordered framework of Er_2CrS_4 denote centered with chromium and erbium, respectively, while blue octahedra denote a disordered array of erbium and chromium cations in the framework of Er_4CrS_7 and $\text{Er}_6\text{Cr}_2\text{S}_{11}$.

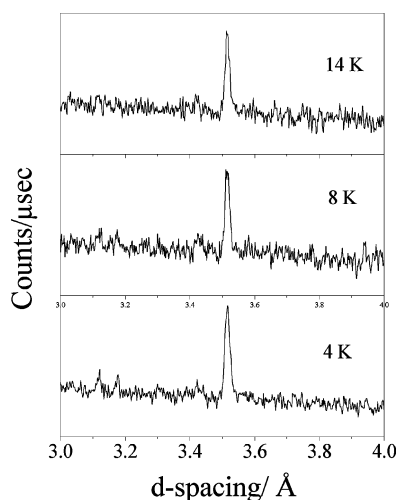


Figure 7. Powder neutron diffraction data collected for $\text{Er}_6\text{Cr}_2\text{S}_{11}$ at 4 K, 8 K, and 14 K, illustrating the disappearance of weak reflections at 3.14 and 3.17 Å on heating through the temperature at which the maximum in the magnetic susceptibility is observed (11.4 K) following subtraction of the contribution due to Er^{3+} .

interaction via an intervening sulfide anion. For the t_{2g}^3 configuration of Cr^{3+} , the $90^\circ d_{\pi}-p_{\pi}/p_{\sigma}-d_{\sigma}$ interaction is predicted to be ferromagnetic.³³ This suggests that the magnetic transition that occurs at 30 K involves ferromagnetic ordering of the chromium ions within the CrS_2^- chains. On further cooling, the weaker interactions associated with the rare-earth ions become significant. This second anomaly at 11 K may therefore be associated with ordering of the erbium sub-lattice. The saturation moment of $7.97(9) \mu_B$ per formula unit at 5 K suggests that it arises from a single uncompensated erbium moment, for which the free-ion value, $g_J J = 9 \mu_B$. This would require the two crystallographically-distinct ferromagnetic chains of chromium-centered octahedra to be antiferromagnetically coupled and the moments associated with one of the erbium sub-lattices to be anti-parallel to that of the other two. Confirmation of this proposed model

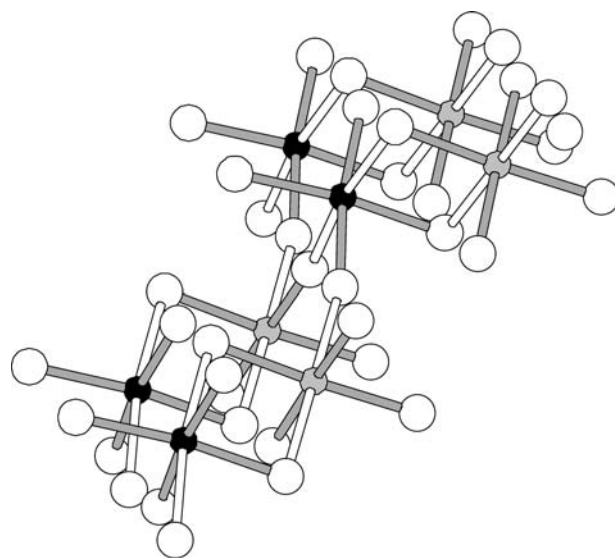


Figure 8. Magnetic exchange pathway due to $\text{Er}^{3+}/\text{Cr}^{2+}$ ordering in the octahedral framework of Er_2CrS_4 . Small solid circles represent $\text{Cr}(5)$, small open circles $\text{Cr}(6)$, and large open circles sulfur. Erbium cations are omitted. Short and long Cr–S distances are indicated respectively by filled and unfilled bonds.

Table 7. Magnetic Properties of Erbium–Chromium Sulfides Derived From Fits to a Curie–Weiss Expression

material	fitting range/ K	$C_{\text{exp}}/\text{cm}^3 \text{ K mol}^{-1}$	$C_{\text{th}}/\text{cm}^3 \text{ K mol}^{-1}$	μ_{eff}/μ_B	μ_{th}/μ_B	θ/K
Er_3CrS_6	80–300	34.32(4)	36.3	8.28(1)	8.52	+8.0(2)
Er_2CrS_4	80–300	27.47(15)	25.95	8.56(3)	8.31	−10.0(1)
Er_4CrS_7	88–300	49.46(13)	48.9	8.90(1)	8.85	−6.4(5)
$\text{Er}_6\text{Cr}_2\text{S}_{11}$	88–300	72.15(16)	74.9	8.49(1)	8.65	+2.4(4)

for the low-temperature magnetically ordered structure requires neutron diffraction measurements.

The structures of the remaining three erbium–chromium sulfides, Er_2CrS_4 , $\text{Er}_6\text{Cr}_2\text{S}_{11}$, and Er_4CrS_7 , investigated here are closely related. Structural relationships of ternary sulfides within the $\text{MnS}-\text{Er}_2\text{S}_3$ and $\text{MnS}-\text{Y}_2\text{S}_3$ pseudo-binary systems^{14,15} have previously been formulated on the basis of chemical twinning,³⁴ in which a homologous series of phases

may be generated from a parent rock-salt type structure through repeated reflection twinning on the {113} planes.^{34,35} While it has been proposed that this description could also be applied to describe ternary phases within the Cr–Er–S system,³⁶ we suggest, on the basis of the crystallographic studies reported here, that the structures of these materials may be better considered as arising from common building blocks. This leads to identification of a structural evolution in these phases that suggest the three erbium–chromium sulfides characterized here may represent the first members of an extended family of materials.

The basic building unit in the structures of Er₂CrS₄, Er₆Cr₂S₁₁, and Er₄CrS₇ is the MS₃ double octahedral chain (Figure 6a) analogous to the isolated MSe₃ chains present in the structure of CeCrSe₃.³⁷ However, in the structures of the erbium–chromium sulfides determined here, these chains are linked through shared edges to form a two-dimensional slab of stoichiometry M₂S₅ (Figure 6b), which serves as the building block for the structures of Er₂CrS₄, Er₄CrS₇, and Er₆Cr₂S₁₁.

In Er₂CrS₄, (Figure 6c) vertex linking of these slabs generates an M₂S₄ framework of linked octahedra. The Cr²⁺ and Er³⁺ ions are ordered within the slabs such that they alternate along the [010] direction within the chains of metal-centered octahedra. The resulting anionic framework of stoichiometry [CrErS₄]³⁻ defines one-dimensional channels directed along [010], in which the seven-coordinate erbium cations, Er(1) and Er(2), that provide the charge-balancing reside. The complete ordering of Cr²⁺ and Er³⁺ in Er₂CrS₄ may be favored by the Jahn–Teller distortion of Cr²⁺:d⁴.

The structure of Er₂CrS₄ is related to that of CaTi₂O₄ (*Cmcm*)³⁸ and its Jahn–Teller distorted variant, CaMn₂O₄ (*Pbcm*).³⁹ On the basis of powder X-ray diffraction, it was suggested that materials Ln₂MS₄ (Ln = Tb–Tm; M = Mg, Mn, Fe)⁹ adopt a structure described in the space-group *Cmcm*, which is closely related to that of CaTi₂O₄. A subsequent single-crystal X-ray diffraction study on non-stoichiometric Yb₂MnS₄ confirms this close-relationship, although the structure was described in *Cmc2*₁.¹² In contrast with the Ln₂MnS₄ structures, in which the rare-earth and the transition-metal atoms are disordered over all the available octahedral sites, in Er₂CrS₄ ordering of erbium and chromium atoms results in a doubling of the *b*-axis, together with a lowering of the symmetry to *P2*₁*ca*. A similar enlargement of the unit cell has been identified, from powder X-ray diffraction data, for other Ln₂CrS₄ (Ln = Y, Ho, Tm, Yb),⁴⁰ suggesting that they are isostructural with Er₂CrS₄. This is supported by a recent powder neutron diffraction study on Y₂CrS₄.⁴¹

The M₂S₅ slab of metal-centered octahedra is also encountered in the structure of Er₄CrS₇ (Figure 6d), which is

isostructural with the binary sulfides Ln₅S₇.^{5,29} The M₂S₅ slabs in Er₄CrS₇ are not fused directly together as occurs in Er₂CrS₄, but are separated by one dimensional MS₄ chains of edge-sharing octahedra, analogous to those present in Er₃CrS₆. However, despite the presence of Cr²⁺:d⁴, the Er³⁺ and Cr²⁺ cations are disordered over the octahedral sites. Refinements indicate that chromium cations show a preference for octahedral sites within the slab, which has a composition (Er_{0.60}Cr_{0.40})₂S₅^{4,8-}, whereas *ca.* 75% of the octahedra in the single-octahedral chains contain erbium cations. The resulting [Er₂CrS₇]⁶⁻ framework, contains channels parallel to the crystallographic *b*-axis, in which are located additional Er³⁺ ions (Er(1)) in a monocapped trigonal prismatic coordination geometry. The shortest cation-cation distances within the slab are *ca.* 3.70 Å, slightly shorter than the intra-chain distances of 3.77 Å. The distribution of rare-earth and transition-metal cations is similar to that in the isostructural Y₄FeS₇^{5,6} and Ho₄FeS₇,⁶ the only two examples of this structure type for which detailed structural characterization has been performed to date.

In the structure of Er₆Cr₂S₁₁ (Figure 6e), pairs of M₂S₅ units are fused to create a double slab. There is disorder of Er³⁺ and Cr²⁺ cations over the available sites within this network of octahedra, with *ca.* 85% of the chromium cations residing in sites within the slab, while the ratio of Er³⁺/Cr²⁺ within the octahedral chains is similar to that in Er₄CrS₇ at 3:1. Successive double slabs are linked by sharing common vertices with single MS₄ octahedral chains to produce a framework of composition [Er₃Cr₂S₁₁]⁹⁻. This network of octahedra contains two types of channel running parallel to the crystallographic *a*-axis. The channels are occupied by six-coordinate Er(1) and seven-coordinate Er(2) cations which serve to balance the charge of the framework. This structure is analogous to that of the binary sulfide Tm₈S₁₁,⁴² while the degree of disorder of Er³⁺ and Cr²⁺ over the octahedral sites is similar to that of Yb³⁺ and Fe²⁺ in Yb₆Fe₂S₁₁.¹⁶ To our knowledge, the Ln₈S₁₁ stoichiometry has not been observed for any other binary rare-earth sulfide. However, the lattice parameters of the ternary compounds Ln₆Mn₂S₁₁ (Ln = Y, Er)^{14,15} and Yb₆Mg₂S₁₁,⁴³ determined by electron diffraction, suggest they crystallize in the same structural type. While the synthesis of the ternary chromium sulfides Ln₆Cr₂S₁₁ (Ln = Gd, Tb, Ho), has been described,¹⁵ structural characterization is yet to be performed.

The presence of M₂S₅ slabs of variable thickness in the structures of each of the ternary phases, Er₂CrS₄, Er₆Cr₂S₁₁, and Er₄CrS₇ suggests a common description can be applied to the structure of these materials. The slabs provide a structural component, [M_{2*n*}S_{4*n*+1}]. This unit shares terminal sulfide ions with MS₄ chains of edge-linked octahedra to

(34) Hyde, B. G.; Andersson, S.; Bakker, M.; Plug, C. M.; O'Keeffe, M. *Prog. Solid State Chem.* **1979**, *12*, 273.

(35) Ferraris, G.; Makovicky, E.; Merlino, S. *Crystallography of Modular Materials*; Oxford University Press: New York, 2004.

(36) Tomas, A.; Guittard, M. *Mat. Res. Bull.* **1980**, *15*, 1547.

(37) Nguyen, H.D.; Etienne, J.; Laruelle, P. *Bull. Soc. Chim. Fr.* **1971**, 2433.

(38) Bertaut, E.F.; Blum, P. *Acta Crystallogr.* **1956**, *9*, 121.

(39) Giesber, H. G.; Pennington, W. T.; Kolis, J. W. *Acta Crystallogr. C* **2001**, *57*, 329.

(40) Tomas, A.; Guittard, M.; Chevalier, R.; Flahaut, J. *C.R. Acad. Sci. Paris, Series C* **1976**, *282*, 587.

(41) Tezuka, K.; Shan, Y. J.; Imoto, H.; Ohoyama, K. *J. Phys. Chem. Solids* **2007**, *68*, 2133.

(42) Zhang, Y.; Franzen, H. F.; Harbrecht, B. *J. Less-Common Met.* **1990**, *166*, 135.

(43) Urones-Garrote, E.; Gómez-Herrero, A.; Landa-Cánovas, A. R.; Otero-Díaz, L. C. *Eur. J. Inorg. Chem.* **2006**, 1436.

generate an anionic framework that may be expressed by the general formula, $[M_{2n+1}S_{4n+3}]^{x-}$, where the charge, x , is determined by the oxidation state of chromium and the Ln/Cr ratio within the framework. The charge balance is provided by additional rare-earth cations which reside at sites of higher coordination number within one-dimensional channels defined by the framework. The anionic network of Er_4CrS_7 and $Er_6Cr_2S_{11}$ can thus be considered the $n = 1$ and $n = 2$ members of a homologous series, while Er_2CrS_4 in which the single octahedral chains are absent is the $n = \infty$ end-member of this family.

The magnetic properties of the $n = 1, 2,$ and ∞ members of this family of ternary phases show significant variation, which can be related to the detailed structure and, in particular, degree of cation ordering. The paramagnetism of Er_4CrS_7 may be traced to the disorder of Er^{3+} and Cr^{2+} over sites in the $Er_3CrS_7^{3-}$ framework. In particular the occupancy of only 40% of the sites in the slab by Cr^{2+} ions may be sufficient to prevent a long-range magnetic exchange pathway from being established. By contrast, in $Er_6Cr_2S_{11}$ where the fraction of octahedral sites occupied by Cr^{2+} in the double slab is only marginally higher at 42%, the degree of coverage appears to exceed the percolation threshold for this topology and magnetic ordering of Cr^{2+} cations appears to occur at low temperatures. Given the cation disorder, we cannot exclude the possibility of a fraction of the erbium ions also being involved in this transition. Further evidence for long-range order below 11.4 K in $Er_6Cr_2S_{11}$ is provided by preliminary powder neutron diffraction data⁴⁴ collected at 4, 8, and 14 K (Figure 7), which reveal that weak reflections present in the 4 K data at 3.14 and 3.17 Å, together with diffuse scattering centered at *ca.* 5 Å, disappear on heating through the temperature of the maximum in $\chi(T)$.

The complete Cr^{2+}/Er^{3+} cation ordering that occurs in Er_2CrS_4 results in a zig-zag magnetic exchange pathway between Cr(5) and Cr(6) ions within the double chains of octahedra that constitute the $ErCrS_5^{5-}$ slab (Figure 8). These interactions are via intervening sulfide anions and are of the 180° superexchange type, involving long and short Cr–S bonds on either side of the anion. The magnitude of the Jahn–Teller distortion ($d_{ax}/d_{eq} \approx 1.14$), suggests that the e_g electron density of Cr^{2+} is located in d_{z^2} orbitals. Therefore,

exchange between Cr(5) and Cr(6) occurs through overlap of a filled d_{z^2} and empty $d_{x^2-y^2}$ with the p-orbital of a sulfide anion. The σ -interaction is predicted to be ferromagnetic in origin, with a moderate antiferromagnetic π -component.³³ There are additional intra-slab 90° antiferromagnetic $d_{\sigma}-p_{\sigma}/p_{\pi}-d_{\pi}$ superexchange interactions between chromium ions located in adjacent double chains. These predictions would suggest that within the $ErCrS_5^{5-}$ slab, chromium ions within a given octahedral chain should be ferromagnetically ordered along the [010] direction, with antiferromagnetic coupling between adjacent chains within the double chain. The double chains would in turn be antiferromagnetically coupled. Inter-slab interactions between chromium cations occur via the erbium-centered octahedra and are therefore likely to be weaker in strength.

In conclusion, we have demonstrated that the magnetic properties of erbium–chromium sulfides are influenced by the degree of rare-earth/transition-metal ordering. In particular on cooling, successive transitions associated with the chromium and erbium sub-lattices are observed in Er_3CrS_6 and Er_2CrS_4 in which cation ordering is complete, suggesting an interaction between the d - and f -electrons of the two types of cation. The cation-disordered phase $Er_6Cr_2S_{11}$ shows magnetic ordering of the chromium sub-lattice only. Characterization of the magnetically-ordered ground states of these materials will require detailed neutron diffraction measurements to be performed as a function of temperature. Significantly, we have shown by single-crystal X-ray diffraction that the structures of Er_2CrS_4 , $Er_6Cr_2S_{11}$, and Er_4CrS_7 may be described in terms of a common building block. This suggests that synthesis of new members of this family of materials, containing thicker slabs of octahedra, may be possible. We are currently pursuing this line of investigation.

Acknowledgment. Financial support from The Royal Society through provision of a Joint Project Grant is gratefully acknowledged. P.V. thanks the U.K. Engineering and Physical Sciences Research Council for the award of an Advanced Research Fellowship and I.S. thanks Heriot-Watt University for a studentship.

Supporting Information Available: Tables of significant bond distances and cif files for each of the single-crystal structure determinations. This material is available free of charge via the Internet at <http://pubs.acs.org>.

IC801482A

(44) Powder neutron diffraction data were collected on the Polaris diffractometer at the ISIS spallation source. The data shown were collected at a scattering angle of 90° in 2θ .